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**3 (Sem-3 /CBCS) PHY HC 2**

**2021**

**(Held in 2022)**

**PHYSICS**

**(Honours)**

Paper : PHY-HC-3026

**(Thermal Physics-II)**

Full Marks : 60

Time : Three hours

**The figures in the margin indicate  
full marks for the questions.**

1. Answer the following questions :  $1 \times 7 = 7$

(a) What is an isobaric process ?

(b) What is the entropy of a perfect crystalline solid at absolute zero temperature ?

Contd.



(c) Whether Maxwell-Boltzmann velocity distribution is applicable to photons.

(d) Joule-Kelvin coefficient of a perfect gas is infinite. (State True or False)

(e) At what temperature, does all molecular motion cease ?

(f) Name the transport phenomenon present in a gas that involves momentum transfer.

(g) How does the diameter of a gas molecule affect mean free path ?

2. Answer the following questions :  $2 \times 4 = 8$

(a) Is temperature a microscopic or macroscopic concept ? Explain.

(b) Differentiate between extensive and intensive variables with examples.

(c) Calculate the average thermal energy of a helium atom at  $27^\circ\text{C}$ .

[ Given  $k_B = 1.38 \times 10^{-23} \text{ m}^2 \text{ kg s}^{-2} \text{ K}^{-1}$  ]

(d) How do viscosity and temperature affect Brownian motion of gas molecules ?

3. Answer any three questions :  $5 \times 3 = 15$

(a) A reversible engine takes in heat from a reservoir of heat at  $527^\circ\text{C}$  and gives out heat to sink at  $127^\circ\text{C}$ . How many calories per second must it take from the reservoir to produce useful mechanical work at the rate of 750 watts ?

(b) Derive an expression for work done during an adiabatic process considering  $n$  moles of an ideal gas.

(c) Explain an experimental method to verify velocity distribution of gas molecules.

(d) The van der Waals constants of oxygen are  $a = 1.382 \text{ L}^2 \text{ bar/mol}$  and  $b = 0.03186 \text{ L/mol}$ . Calculate its Boyle's temperature and temperature of inversion.  $2\frac{1}{2} + 2\frac{1}{2} = 5$

(e) Derive Clausius-Clapeyron equation.

4. Answer the following questions :

$10 \times 3 = 30$

(a) Using Maxwell's thermodynamic relations, derive  $T_{ds}$  equations. 10



**Or**

What is Gibbs free energy ? Using Gibbs free energy  $G$ , show that

$$G = -T^2 \left[ \frac{\partial}{\partial T} \left( \frac{G}{T} \right) \right]_P$$

where the symbols have their usual meanings. 1+9=10

- (b) Define coefficient of diffusion. Discuss the theory of diffusion in a gas and show that coefficient of diffusion is directly proportional to square root of temperature.

1+2+7=10

**Or**

Derive the van der Waals equation of state and calculate the value of critical constants. 5+5=10

- (c) What do you mean by thermodynamic scale of temperature? Show that the thermodynamic scale of temperature is identical with the perfect gas scale of temperature. 3+7=10

**Or**

Write short notes on the following : (**any two**) 5×2=10

- (i) Carnot cycle
- (ii) Degrees of freedom
- (iii) Joule-Thomson cooling